

Lecture in line with the course  
„Biotechnology of trees and fungi“

# Cellulose - Hemicellulose

Dr. Christian Schöpfer

Phone: 0551 - 39 9745

mailto: [cschoep@gwdg.de](mailto:cschoep@gwdg.de)

# Major cell wall polysaccharides

⇒ Celluloses

⇒ Hemicelluloses

⇒ Pectic polysaccharides

# Cell wall components

## ⇒ Microfibrillar structure

⇒ Celluloses ( $\beta$ -1,4 glucan)

## ⇒ Matrix structure

⇒ Pectins (galactan, arabinan, etc.)

⇒ Hemicelluloses (xylan, mannan, etc.)

⇒ Proteins (arabinogalactan, enzymes, etc.)

⇒ Phenolics (lignin, ferulic acid, etc.)

# Polysaccharide composition (%) in different cell wall layers

	Middle lamella, primary wall	Secondary wall S <sub>1</sub>	Secondary wall S <sub>2</sub> outer part	Secondary wall S <sub>2</sub> inner part
<b><u>Birch</u></b>				
Galactan	16,9	1,2	0,7	0,0
<b>Cellulose</b>	<b>41,8</b>	<b>48,8</b>	<b>49,4</b>	<b>60,0</b>
Glucomannan	3,1	2,8	2,1	5,1
Arabinan	13,4	1,9	1,5	0,0
<b><u>Spruce</u></b>				
Galactose	16,3	8,0	0,0	0,0
<b>Cellulose</b>	<b>33,4</b>	<b>55,2</b>	<b>63,6</b>	<b>64,3</b>
Glucomannan	7,9	18,1	24,4	23,7
Arabinan	29,3	1,1	0,8	0,0

# Chemical composition of Japanese soft- and hardwoods

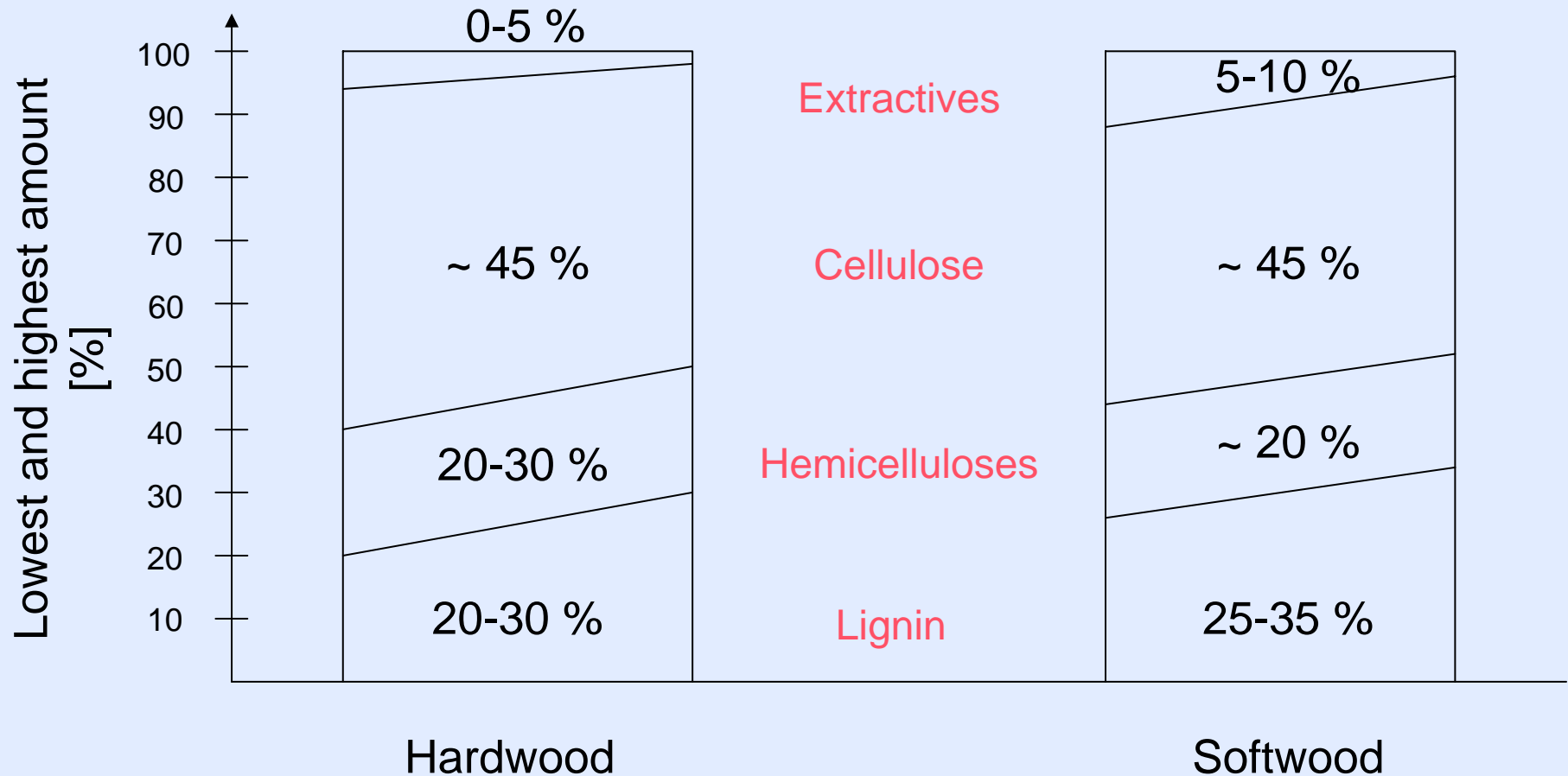
	<b><math>\alpha</math>- cellulose (%)</b>	<b>Pentosan (%)</b>	<b>Mannan (%)</b>	<b>Galactan (%)</b>	<b>Proteins (%)</b>
<i>Pinus densiflora</i>	<b>31.0 - 41.3</b>	9.9 - 12.9	3.5 - 10.1	0.3 - 0.8	0.8
<i>Abies firma</i>	<b>38.2</b>	8.5 - 10.9	7.9	1.3	0.9
<i>Larix kaempferi</i>	<b>30.6 - 38.9</b>	4.8 - 11.8	4.1 - 7.8	0.6 - 9.0	0.3 - 0.5
<i>Fagus crenata</i>	<b>40.7 - 45.9</b>	21.3 - 26.9	0.0	0.1 - 1.6	0.4 - 1.2
<i>Quercus serrata</i>	<b>37.1 - 43.5</b>	18.3 - 24.0	0.0	0.4 - 1.8	0.7 - 1.0
<i>Acer mono</i>	<b>37.1 - 48.9</b>	18.7 - 25.7	0.0	0.3 - 1.5	0.6 - 0.7

# Amount of different sugars in wood

	1 Galactose (%)	2 Glucose (%)	3 Mannose (%)	4 Arabinose (%)	5 Xylose (%)
Average amount in <b>softwood</b>	9.3	64.5	12.0	3.3	10.3
Average amount in <b>hardwood</b>	2.3	65.2	2.6	1.5	28.3

1 - 3 = hexoses, 4 and 5 = pentoses

# Cellulose and hemicellulose amount in softwood and hardwood



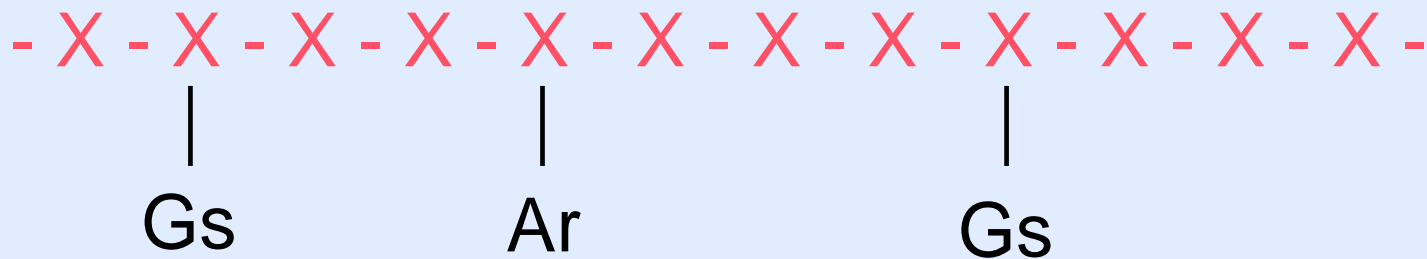
# Characteristics of hemicellulose

- ⇒ The term hemicellulose unites all carbohydrates of wood that can be easily hydrolysed
- ⇒ Built up by different glycosidic complex chain molecules that are linked to pentose and hexose
- ⇒ Hemicelluloses feature an average degree of polymerisation of 120 - 200
- ⇒ Next to hexose and pentose further units like uronic acids are contained in hemicelluloses



# Assembling units of hemicelluloses - softwood

## “Softwood-xylan”

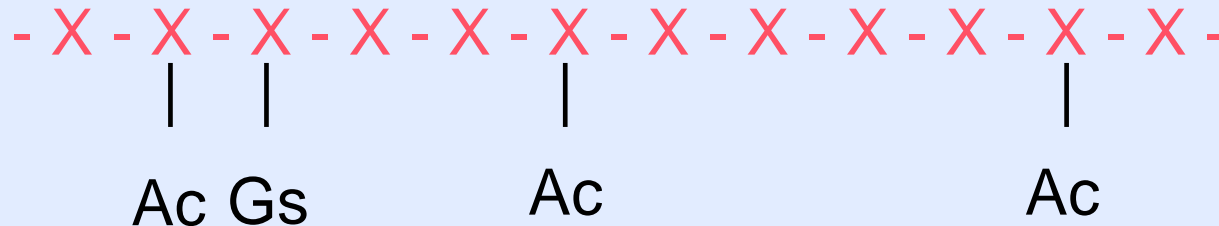


X = Xylose, Gs = Glucuronic acid, Ar = Arabinose



# Assembling units of hemicelluloses - hardwood

“Hardwood-xylan”

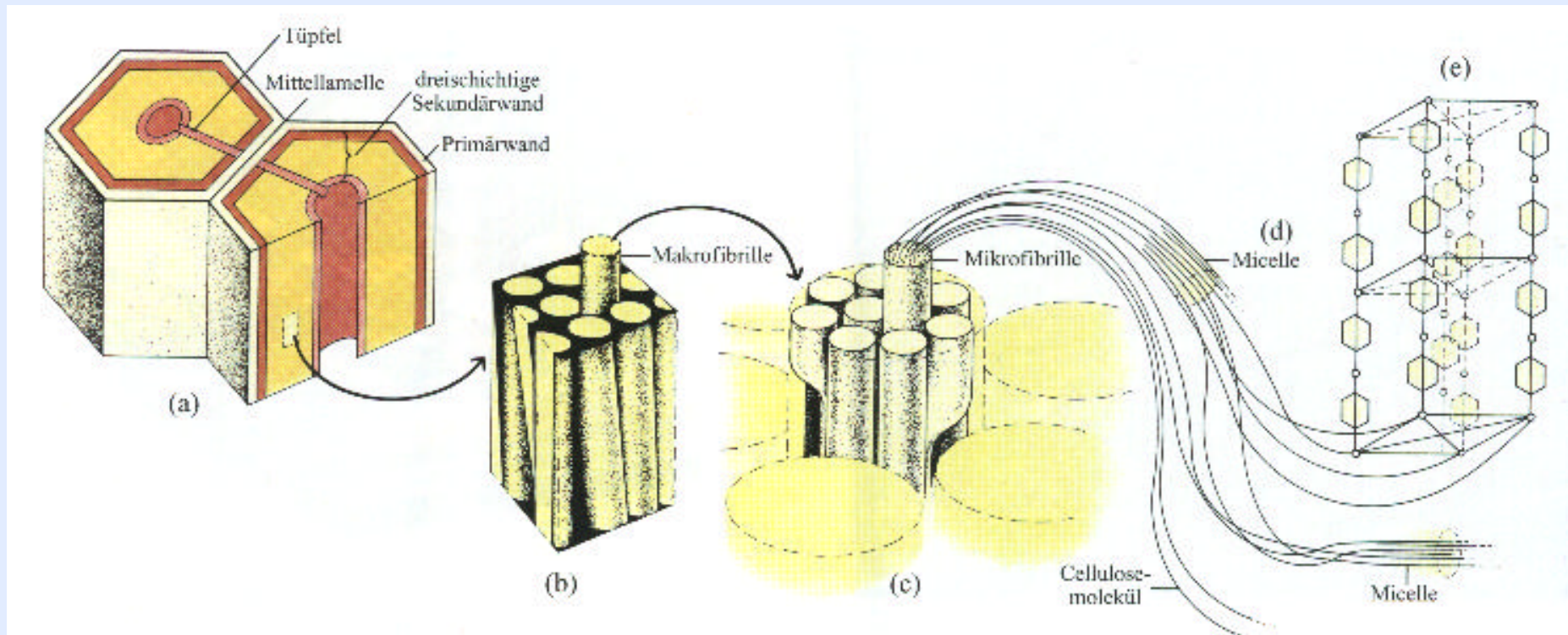


X = Xylose, Gs = Glucuronic acid, Ac = Acetyl

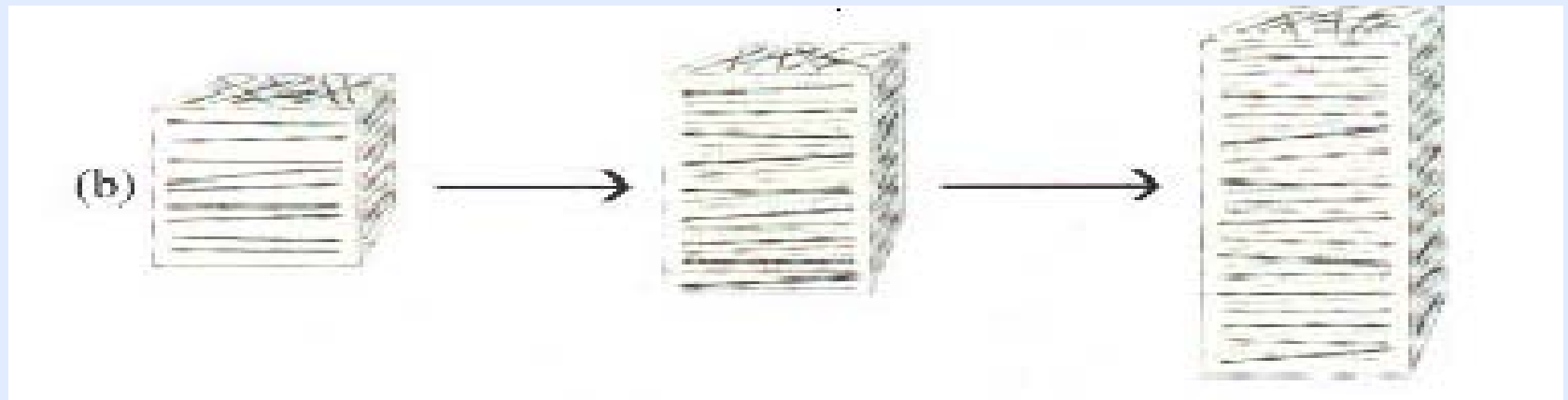
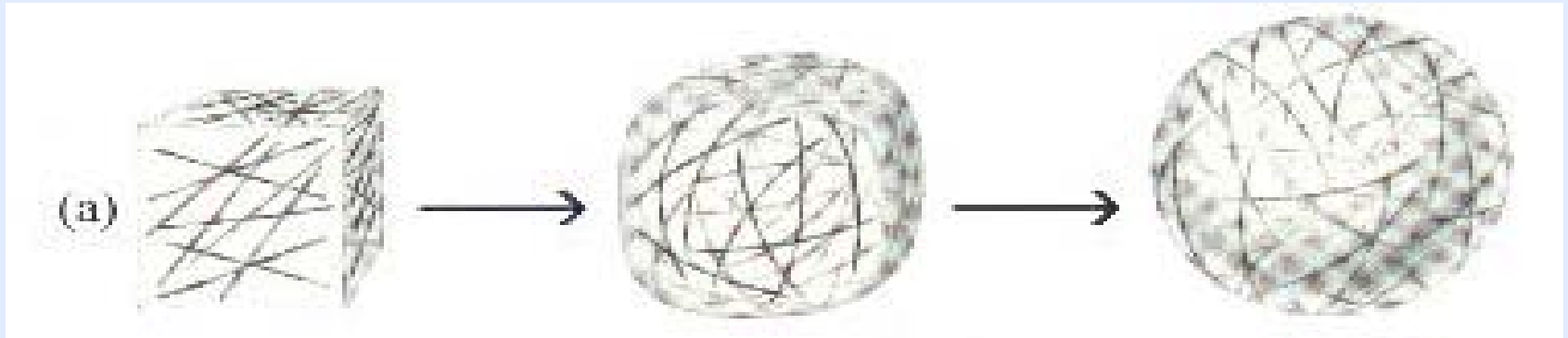
# Characteristics of cellulose

- ⇒ Built up exclusively by  $\beta$ -D-glucose units
- ⇒  $\beta$ -D-glucose units are linked to high molecular chain molecules
- ⇒ The average degree of polymerisation of cellulose in a native state is 10.000 to 15.000
- ⇒ A fast growing tree produces approximately 14 g of cellulose per day
- ⇒ Cellulose is the most important produced biomass worldwide!

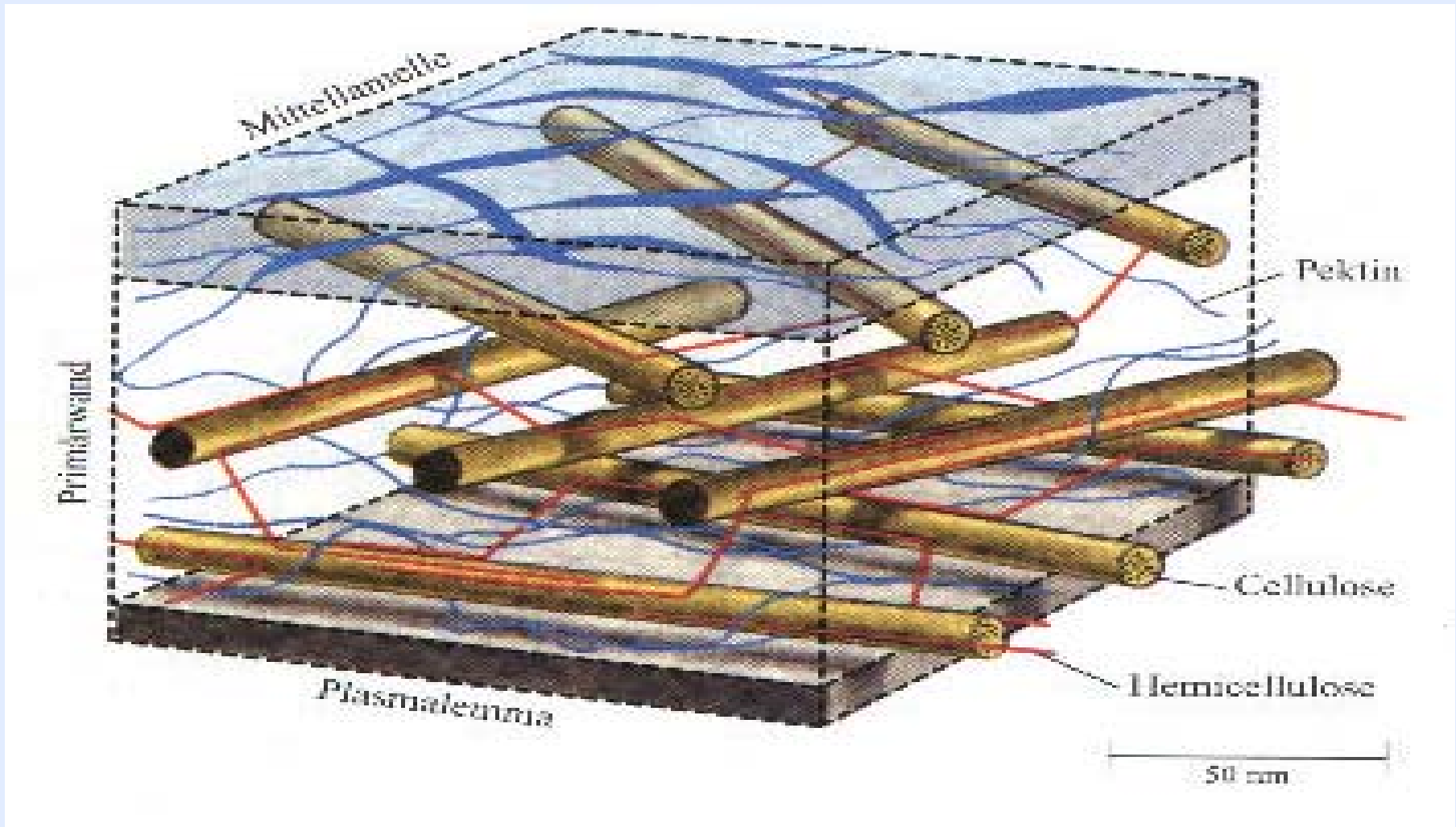
# Appearance of cellulose



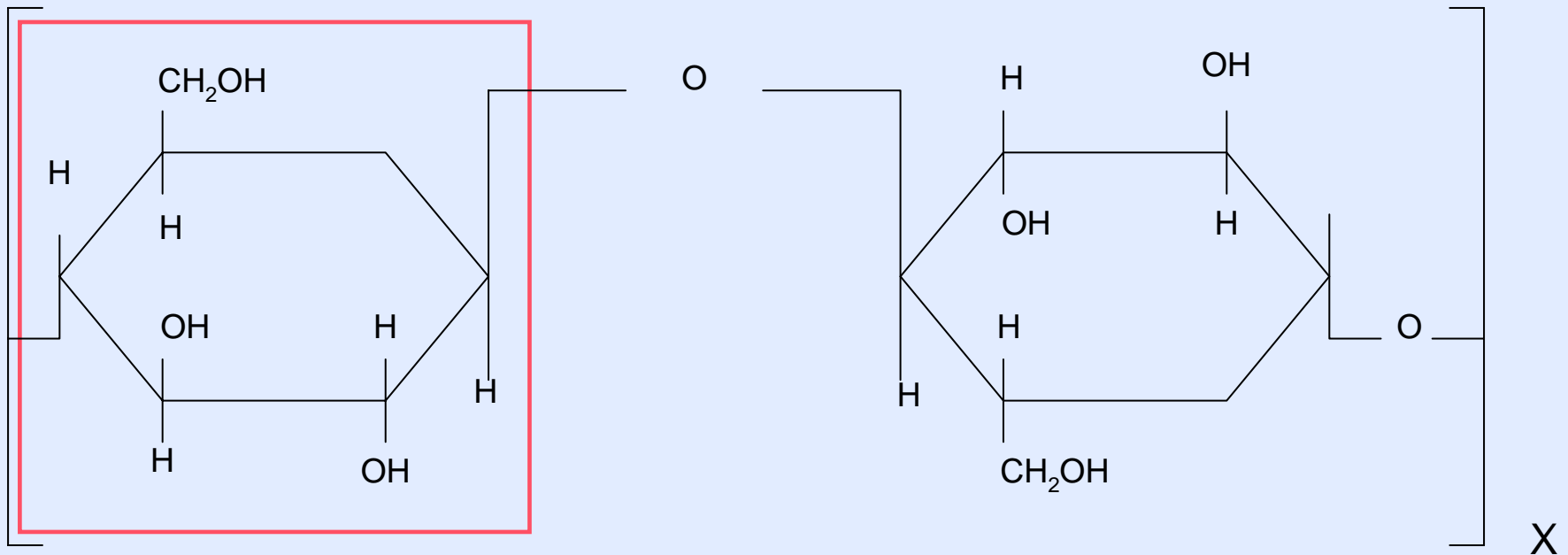
# Orientation of cellulose



# Cellulose formation



# Chemical structure of cellulose

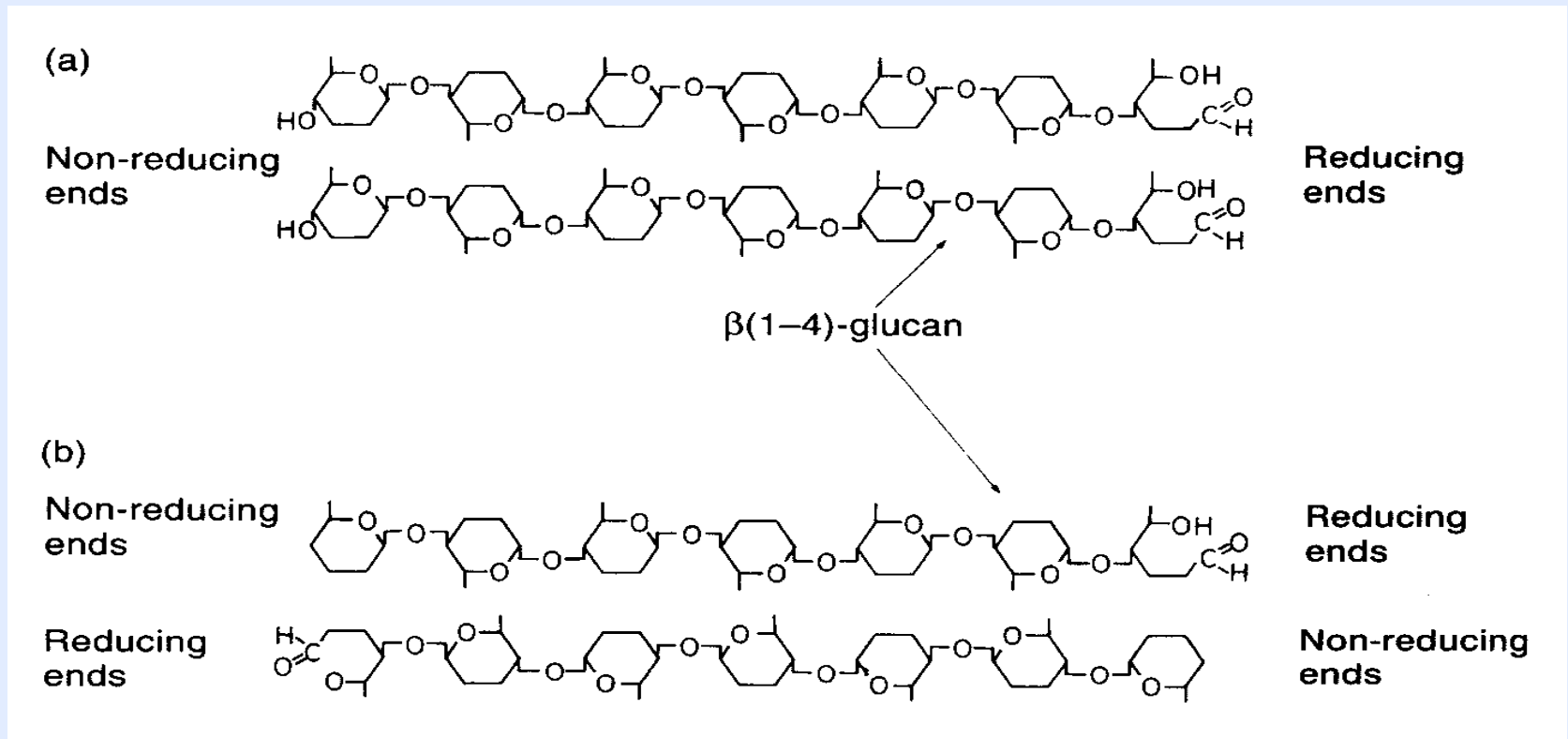


$\beta$ -D-glucose unit

X = degree of polymerisation



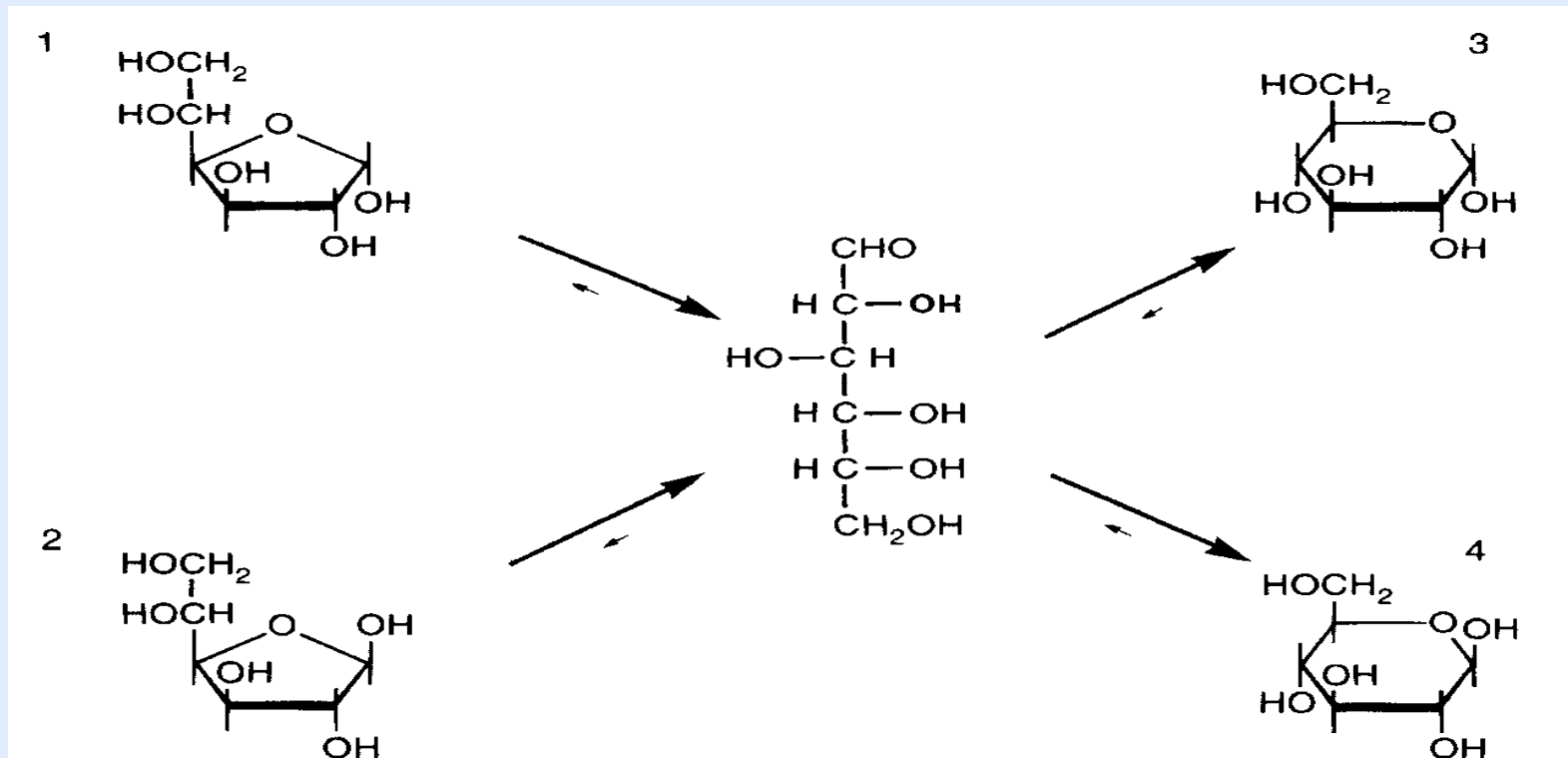
# Cellulose in crystalline structure



(a): parallel configuration

(b): antiparallel configuration

# Furanose - pyranose



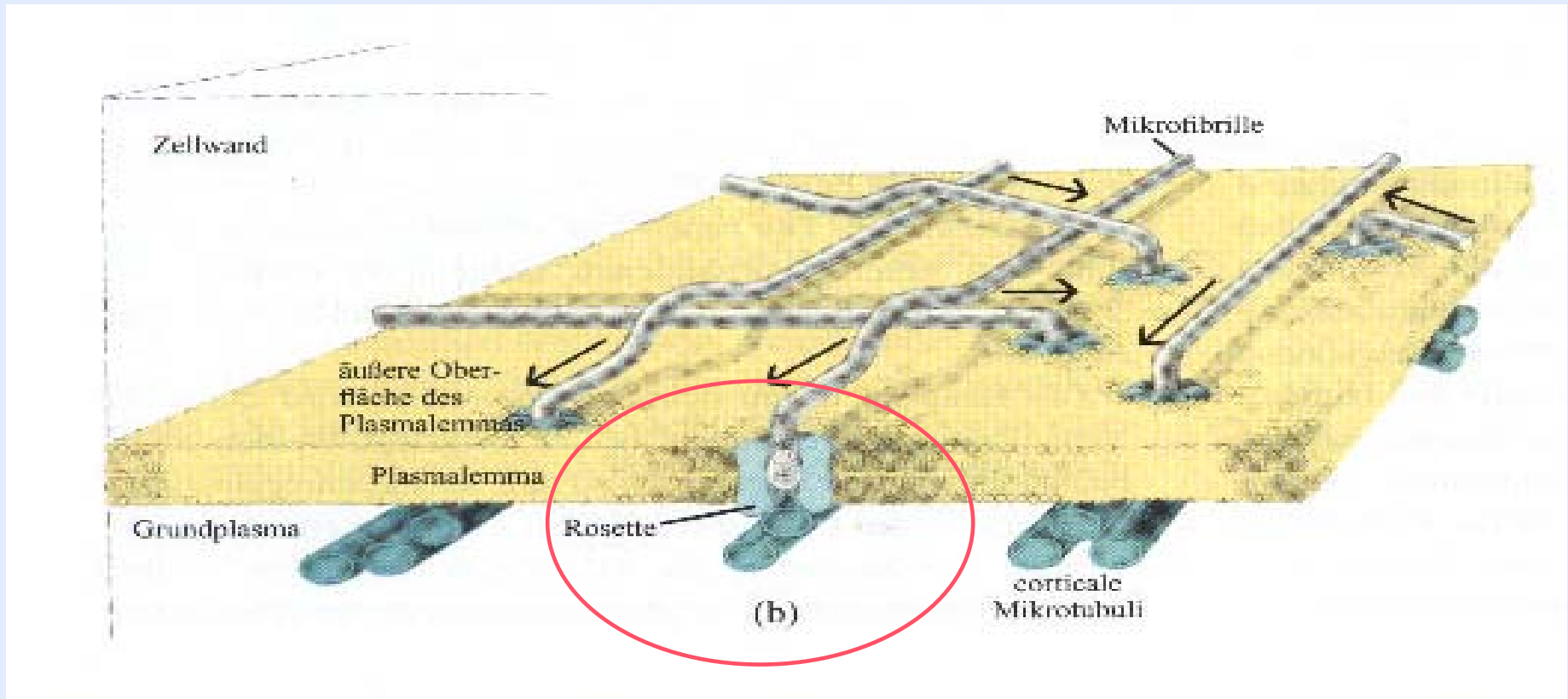
(1):  $\alpha$ -D-glucofuranose (< 1%)

(3):  $\alpha$ -D-glucopyranose (36 %)

(2):  $\beta$ -D-glucofuranose (< 1%)

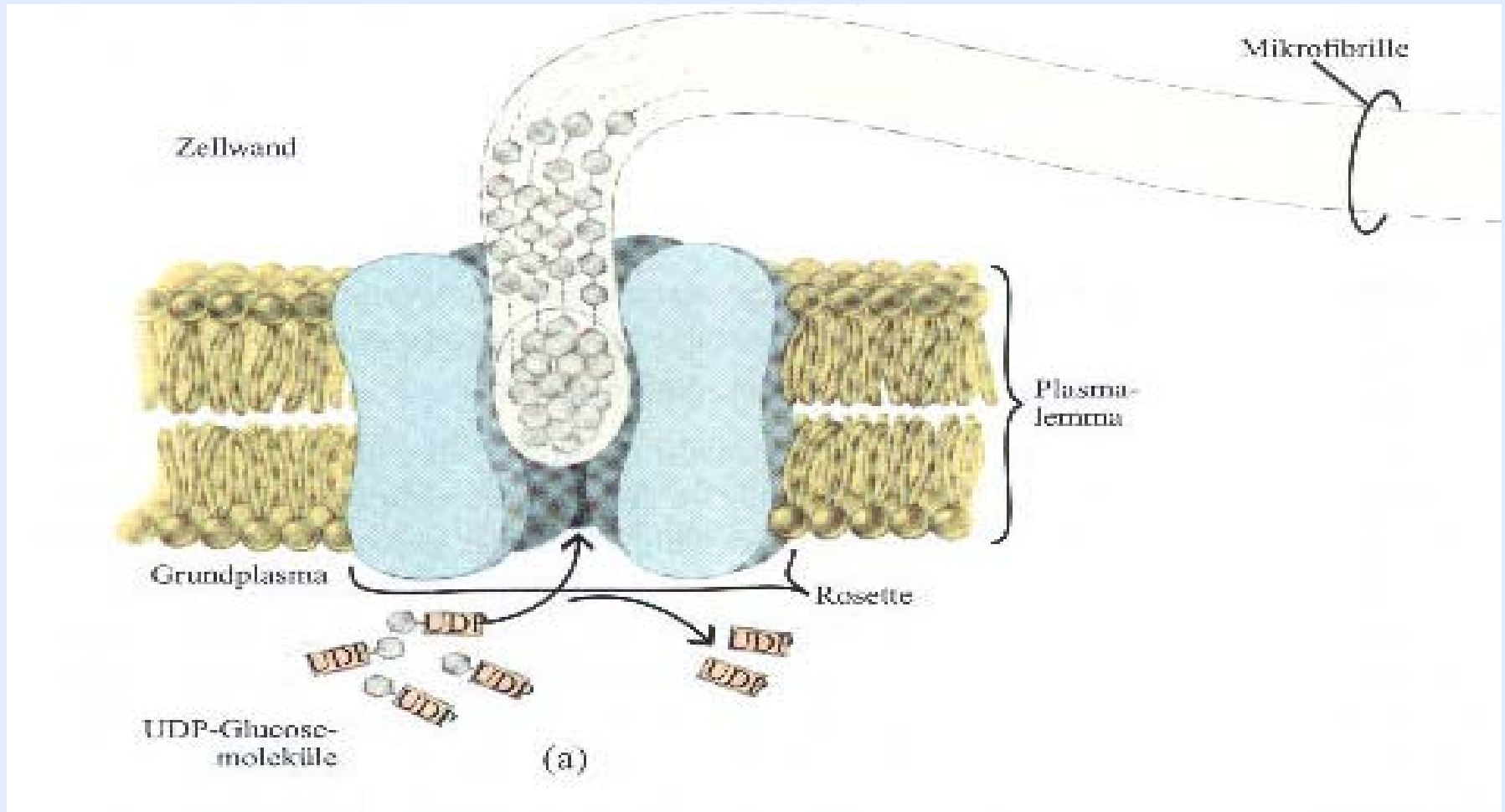
(4):  $\beta$ -D-glucopyranose (64 %)

# Cellulose synthesis I



Cellulose synthetase is located in rosettes in the plasma membrane using UDP glucose as substrate for polymerisation

# Cellulose synthesis II



# Determination of holocellulose

1. Weigh in 2 g of extractive free wood and put in 100 ml Erlenmeyer and add 80 ml water
2. Add 0.25 ml acetic acid (100 %), 0.75 g  $\text{NaClO}_2$  and close Erlenmeyer with a bubble or a cover
3. Put in hot water bath at 80°C for 1 h
4. Repeat steps from 2 and 3 three times for hardwoods and four times for softwoods
5. Cool the samples in ice water bath and filter the samples on a glass filter in a vacuum suction
6. Wash the samples with 100 ml ice water and than with 25 ml acetone
7. Put the samples on oven dried aluminium cups and dry at 105 +/- 3° C and than weigh the samples

Wise, L. E.; Murphy, M., D'Addieco, A. A. 1946: Chlorite holocellulose, its fractionation and bearing on summative wood analysis and on studies on the hemicelluloses. Paper Trade J. 122:11-19

# Literature

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